

# Gerda meeting

LNGS 26-28/06/2006

Report on ICP-MS measurement  
carried out at LNGS  
on different GERDA samples

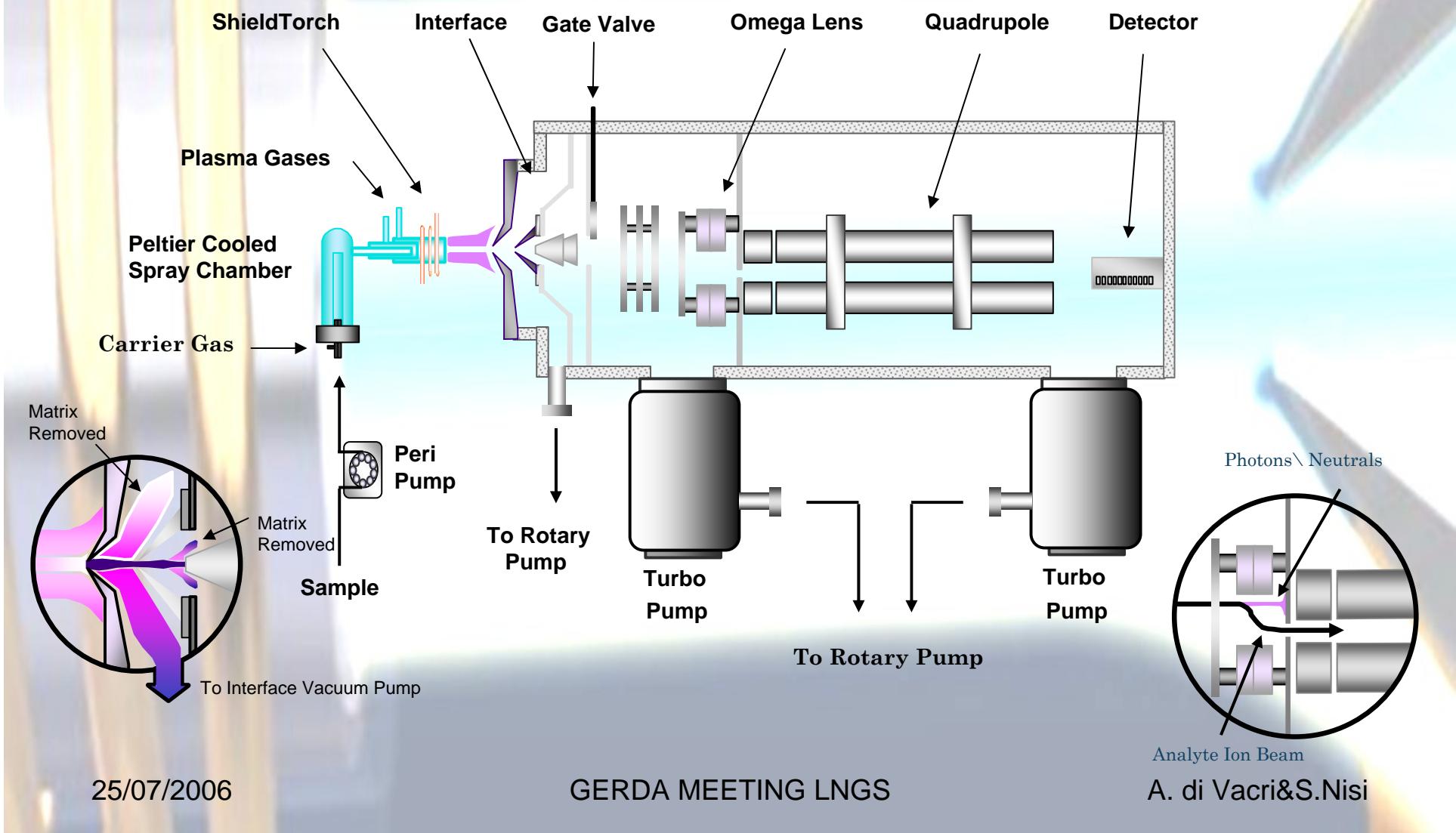
LNGS Chemistry Service & Chemical Plant

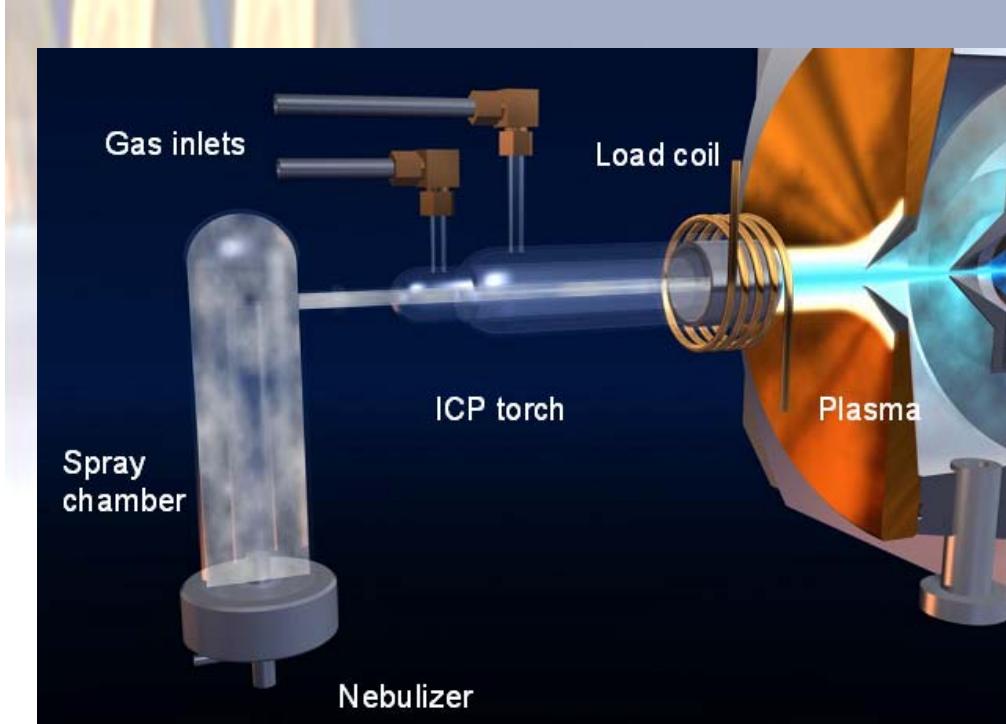
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# What is ICP-MS ?

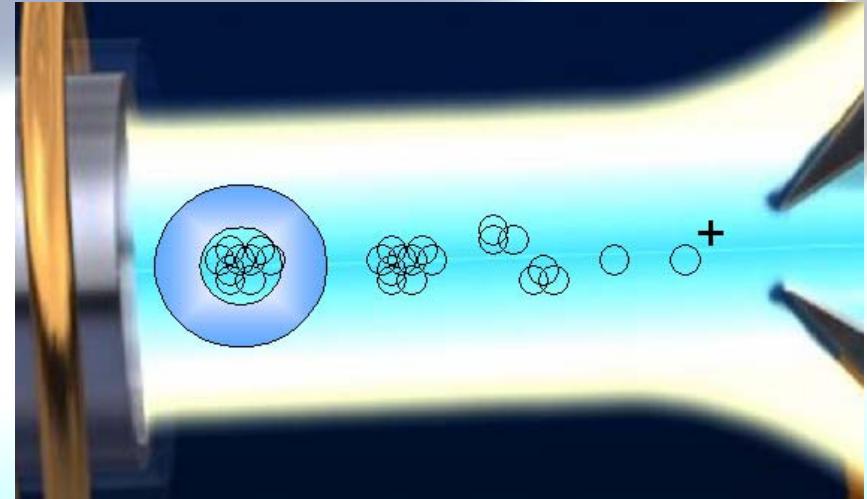
- An inorganic (multi-elemental) analytical technique
- ICP - Inductively Coupled Plasma
  - high temperature ion source decompose and ionize all elements present in the sample
- MS - Mass Spectrometer
  - quadrupole scanning spectrometer
  - mass range from 3 to 260 amu (Li to U...)
  - separates all ions in rapid sequential scan
  - ions measured using dual mode detector
    - ppt to ppm levels
    - isotopic information available

# Agilent Technologies 7500 Series ICP-MS Schematic

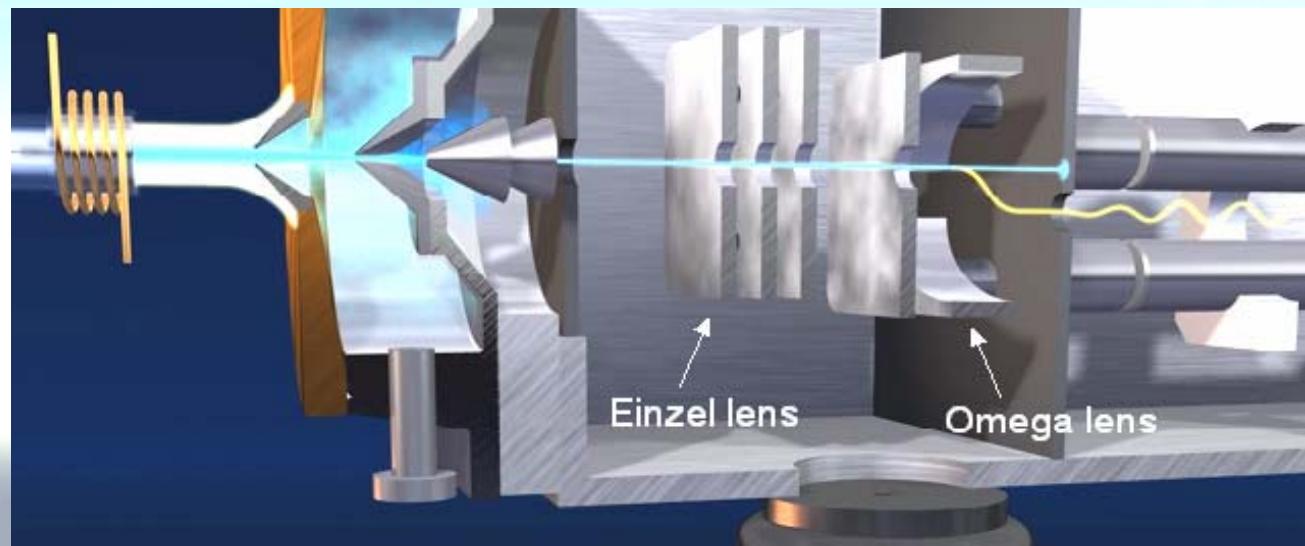




## ICP MS Agilent 7500a:



- ◆ Sample introduction;
- ◆ Aerosol drying, Atomization, Ionization;
- ◆ Analysis;
- ◆ Detection.

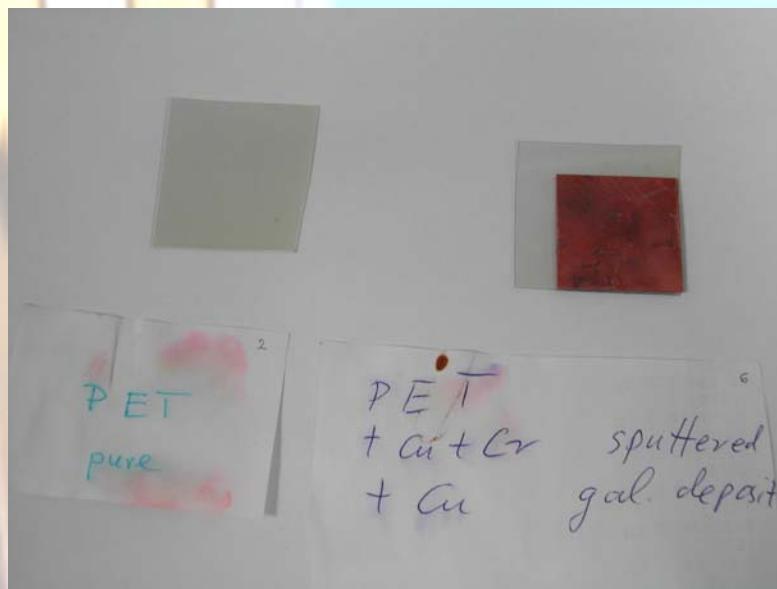


# ICP-MS potentiality&sensitivity depending on matrix type

	"cleaned" Solution	Metallic sample	Plastic sample
Ideal sample amount	10ml	0,1-0,5g	0,5-5g
Element	ppb	ppb	ppb
$^{39}\text{K}$	25	2500	250
$^{208}\text{Pb}$	0,005	0,5	0,1
$^{232}\text{Th}$	0,001	0,1	0,01
$^{238}\text{U}$	0,001	0,1	0,01

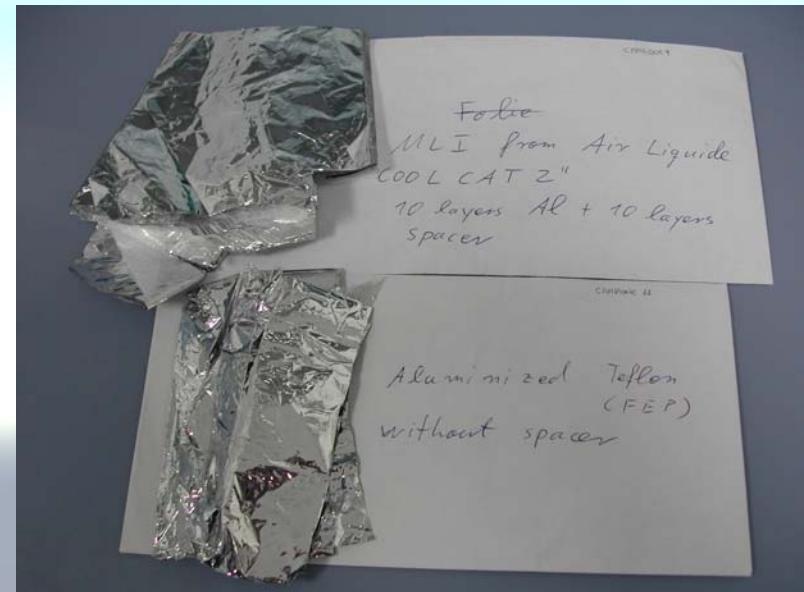
# ICP-MS measurements for GERDA

*Purpose:* evaluate the K, Pb, Th and U contamination in polymeric substrates with and without metallic cladding by means of ICP-MS



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sample #	Sample description
<b>PLASTIC FOILS</b>	
1	Pure PEN (PolyEthylene-Naphthalate)
2	Pure PET (PolyEthylene-Terephthalate)
<b>MULTI-LAYER SAMPLES</b>	
3	PEN + (Cr+Cu sputtered) + (Cu galvanic deposit) + (PCB processing)
4	PET + (Cr+Cu sputtered) + (Cu galvanic deposit) + (PCB processing)
5	PEN + (Cr+Cu sputtered) + (Cu galvanic deposit)
6	PET + (Cr+Cu sputtered) + (Cu galvanic deposit)
7	PEN + (Cr+Cu sputtered)
8	PET + (Cr+Cu sputtered)
<b>SUPER-INSULATION MATERIALS</b>	
9	"COOL CAT 2" from Air Liquid, 10 layers Al+10 layer spacer
10	NAC-2 foil Polyester spacer "bonded" with Al
11	Aluminized Teflon (FEP) without spacer

# Before starting trace analysis: choose of reagents...

	K [ppb]	Pb [ppt]	Th [ppt]	U [ppt]
HNO <sub>3</sub> UP 20%	<25	<10	<1	<1
HNO <sub>3</sub> SP 20%	<25	15	<1	<1
HCl UP 20%	<25	20	<1	<1
Detergent ELMA65 1%	<25	120	1	13
DETERGENT8 ALCONOX 1%	250	5	<0.2	<0.2



- ✓ Ultra-pure acids selected for all steps of the sample preparation;
- ✓ Detergent8 fine for the purpose even if K level is quite high;

# ... and containers: polyethylene bags for washing

	K [ppb]	Pb [ppt]	Th [ppt]	U [ppt]
PE bag H <sub>2</sub> O demi rinse HNO <sub>3</sub> UP 2%	<25	8	<1	<1

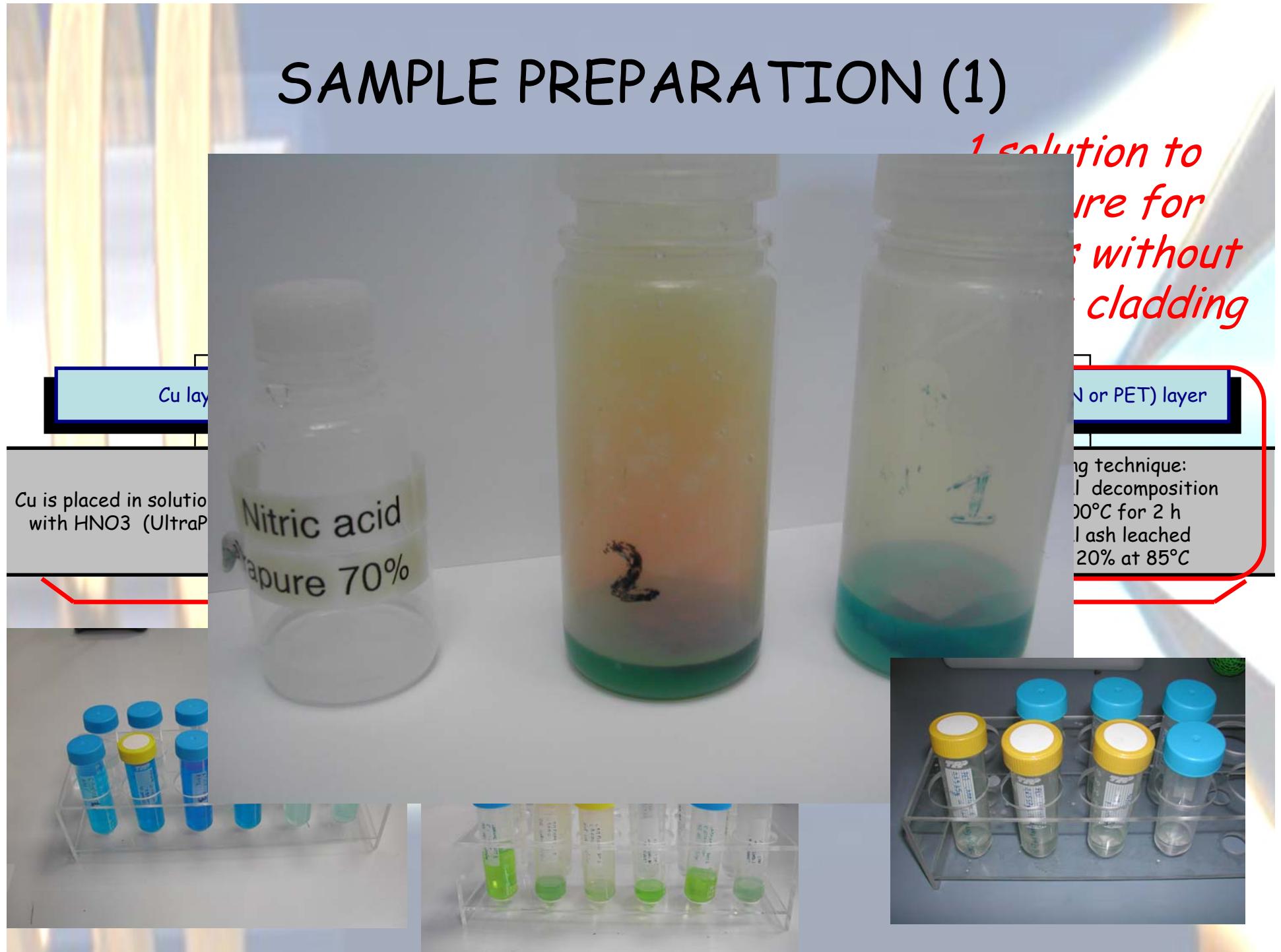
# and crucibles for ashing

	K [ppb]	Pb [ppt]	Th [ppt]	U [ppt]
Pt crucible *	<25	1300	30	28
Quartz crucibles 1 *	<25	310	~1	<1
Quartz crucibles 2 *	<25	59	<1	<1
Porcelain crucible 1 *	<25	130	6	2
Porcelain crucible 2 *	<25	27	3	1
Quartz crucible HNO <sub>3</sub> UP 1% 5 days	<25	<5	<1	<1

contamination level is the lowest also after high T thermal cycle

\* 2h muffle furnace 600 °C + HNO<sub>3</sub> 20% 80 °C 30 min  
→ procedure simulation

# SAMPLE PREPARATION (1)



# SAMPLE PREPARATION (2)

## Super-insulation material

1) weighed by analytical balance;

2) cut;

3) weighed by ultrasound bath (UP water)

## Al layer

Al is placed in solution by acid digestion  
with  $\text{HNO}_3$  (UP)

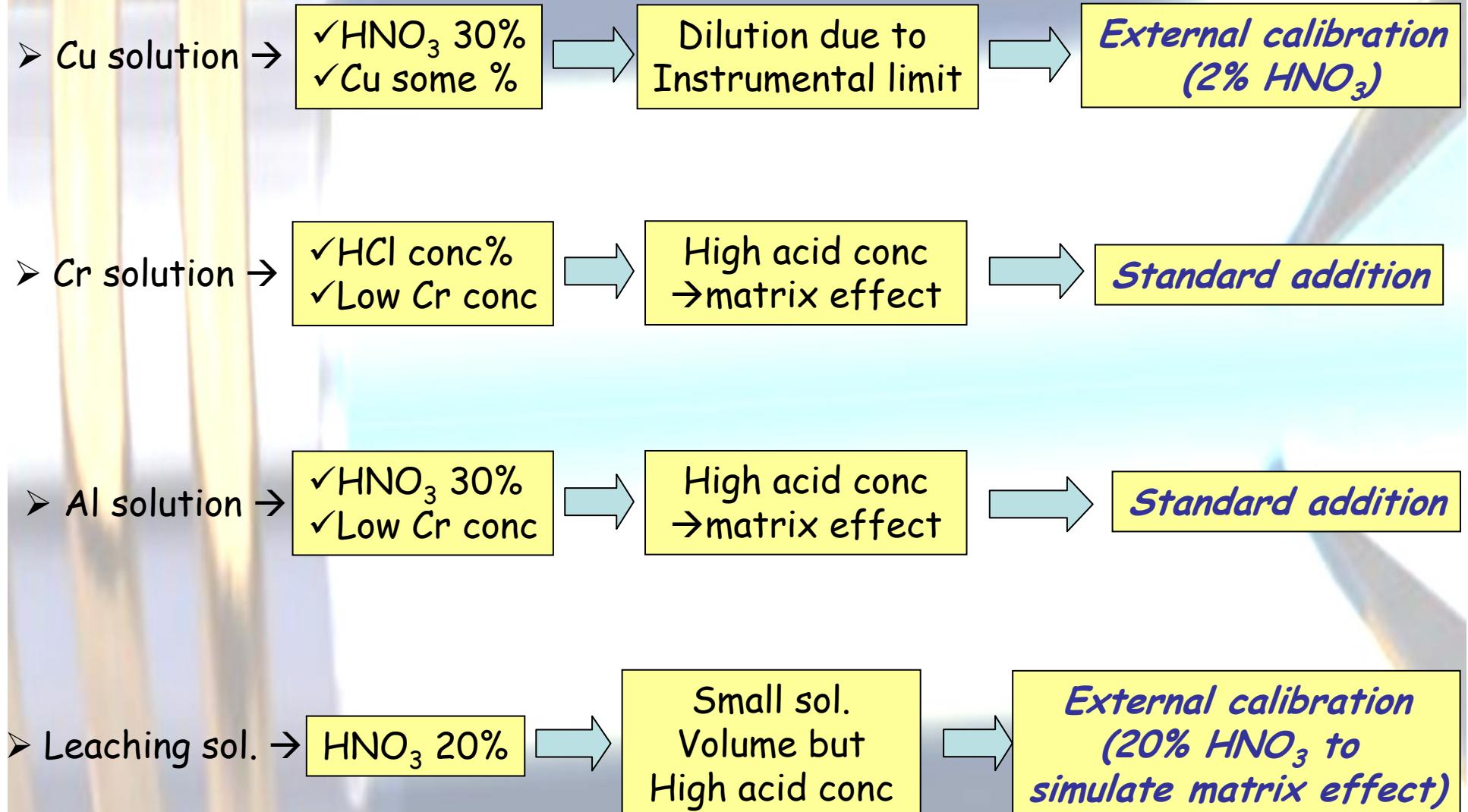
## Plastic layer

Dry ashing technique:

- 1) combustible portion destroyed by thermal decomposition up to  $600^\circ\text{C}$  for 2 hours
- 2) residual ash dissolved in  $\text{HNO}_3$  20% at  $85^\circ\text{C}$  repeated twice

*2 solutions to measure*

# Sample treatment



# Collected solutions for ICP-MS & AAS



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# Data analysis

SAMPLE 4				
	K [ppb]	Pb [ppt]	Th [ppt]	U [ppt]
Washing sol. 1%	$174 \pm 1\%$	$60 \pm 2\%$	<1	<1
BLK for etching	<25	<5	<1	<1
Cu	$17 \pm 3\%$	$201 \pm 2\%$	<1	<1
Cr	<25	$302 \pm 22\%$	$2 +/- 25\%$	$2 \pm 25\%$
BLK for ashing	<25	$91 \pm 4\%$	<1	<1
PET	<25	$1592 \pm 6\%$	$19 \pm 5\%$	$111 \pm 7\%$

BLK subtraction and dilution factor (Al & Cr weight by AAS)

S 4	Weight [ % ]	K [ppb]	Pb [ppb]	Th [ppt]	U [ppt]
Cu	70.58	$3014 \pm 2.5\%$	$35 \pm 1.6\%$	< 174	< 174
Cr	0.32	<144200	$1743 \pm 22\%$	$11540 \pm 25\%$	$11540 \pm 25\%$
PET	29.1	<560	$35.5 \pm 6\%$	$433 \pm 5\%$	$2470 \pm 7\%$

Taking into account mass fraction of each layer respect whole sample...

## polymeric substrates w/o metallic cladding

- ✓ PEN samples cleaner than PET ones.
- ✓ Th & U contribution of Cu and Cr layers under detection limit.
- ✓ In multi-layer samples with a PET core contamination mainly comes from the plastic and there is an indication that during the etching of the Cr layer some contaminants have been extracted from the plastic core.

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		<b>Weight</b> [ % ]	<b>K</b> [ppb]	<b>Pb</b> [ppb]	<b>Th</b> [ppt]	<b>U</b> [ppt]
1	Pure PEN	100	< 393	26 ± 3	< 16	< 16
2	Pure PET	100	480 ± 110	53 ± 5	522 ± 52	3714 ± 37
3	Cu	77.59	3210 ± 160	16.3 ± 0.1	< 124	134 ± 2
	Cr	0.21	< 183	1.4 ± 0.2	< 7.3 ± 3.7	< 7
	PEN	16.2	< 66	4.2 ± 0.4	< 3	< 3
	Whole samp.		3210 - 160 <sup>+170</sup>	21.9 ± 0.5	7 - 4 <sup>+42</sup>	134 - 2 <sup>+3</sup>
4	Cu	70.58	2127 ± 53	24.7 ± 0.4	< 123	< 123
	Cr	0.32	< 461	5.6 ± 1.2	37 ± 9	37 ± 9
	PET	29.1	< 162	10.3 ± 0.6	126 ± 6	719 ± 50
	Whole samp.		2130 - 50 <sup>+170</sup>	40.6 ± 1.4	163 - 11 <sup>+42</sup>	756 - 51 <sup>+65</sup>
5	Cu	82.03	1646 ± 21	7.0 ± 0.4	< 135	< 135
	Cr	0.25	< 270	3.5 ± 0.2	< 10	< 10
	PEN	17.72	< 85	4.3 ± 0.3	< 3	< 3
	Whole samp.		1646 - 21 <sup>+97</sup>	14.8 ± 0.5	< 148	< 148
6	Cu	88.26	< 1475	5.9 ± 0.4	< 147	< 147
	Cr	0.17	< 298	3.8 ± 0.4	61 ± 12	19 ± 1
	PET	11.57	84.2 ± 3	7.8 ± 0.4	64.8 ± 8	586 ± 18
	Whole samp.		84 - 3 <sup>+500</sup>	17.5 ± 0.4	126 - 14 <sup>+51</sup>	605 - 18 <sup>+52</sup>
7	Cu	17.29	1290 ± 170	8.6 ± 0.4	< 76	< 76
	Cr	1.68	< 3041	10.4 ± 0.5	< 122	< 122
	PEN	81.03	< 365	55 ± 3	< 144	< 90
	Whole samp.		1290 - 170 <sup>+1040</sup>	74 ± 3	< 342	< 288
8	Cu	16.93	1574 ± 79	20.7 ± 0.3	< 85	86 ± 14
	Cr	1.27	< 3430	26 ± 2	< 140	< 140
	PET	81.8	412 ± 29	45 ± 2	452 ± 27	3700 ± 190
	Whole samp.		1990 - 80 <sup>+1150</sup>	92 ± 3	452 - 84 <sup>+61</sup>	3790 ± 190

# Super-insulation materials

#		Weight [ % ]	K [ppb]	Pb [ppb]	Th [ppt]	U [ppt]
9	Al	0.73	526 ± 32	122 ± 2	44 ± 4	23 ± 4
	MLI	99.27	3860 ± 120	158 ± 6	77 ± 5	92 ± 6
	MLI 2nd		51 ± 4	5.5 ± 0.2	2.8 ± 0.4	< 2.3
	Whole		4435 ± 120	286 ± 6	124 ± 6	115 ± 7
10	Al	0.55	1045 ± 105	545 ± 11	831 ± 50	458 ± 9
	NAC-2	99.45	1775 ± 71	288 ± 12	864 ± 43	1420 ± 71
	NAC-2 2nd		< 270	< 10	84 ± 6	25 ± 5
	Whole		2820 - <sub>130</sub> <sup>160</sup>	833 - <sub>16</sub> <sup>17</sup>	1779 ± 66	1903 ± 72
11	Al	1	824 ± 16	138 ± 6	90 ± 7	91 ± 4
	Teflon	99	306 ± 15	13.6 ± 0.8	161 ± 10	23 ± 2
	Teflon 2nd		949 ± 19	4.2 ± 0.1	22 ± 2	4.7 ± 0.3
	Whole		2079 ± 29	156 ± 6	273 ± 12	119 ± 4

Sample 9 ("COOL CAT 2" from Air Liquid) and sample 11 (Aluminized Teflon, FEP) are cleaner than sample 10 (NAC-2).

# Specific activities for $^{40}\text{K}$ , $^{232}\text{Th}$ , $^{238}\text{U}$ deriving from the concentrations obtained by ICP-MS

Sample	$^{40}\text{K}$ [mBq/kg]	$^{232}\text{Th}$ [mBq/kg]	$^{238}\text{U}$ [mBq/kg]
1	< 12	< 0.07	< 0.2
2	$14.6 \pm 3.3$	$2.1 \pm 0.2$	$46.1 \pm 0.5$
3	$97 \pm 5$	$0.03_{-0.02}^{+0.17}$	$1.66_{-0.02}^{+0.04}$
4	$65_{-2}^{+5}$	$0.66_{-0.04}^{+0.17}$	$9.4_{-0.6}^{+0.8}$
5	$49.9_{-0.6}^{+2.9}$	< 0.6	< 1.8
6	$2.6_{-0.1}^{+15.2}$	$0.51_{-0.06}^{+0.21}$	$7.5_{-0.2}^{+6.4}$
7	$39_{-5}^{+35}$	< 1.4	< 3.6
8	$60_{-2}^{+35}$	$1.8_{-0.3}^{+0.2}$	$47 \pm 2$
9	$135 \pm 4$	$0.50 \pm 0.02$	$1.43 \pm 0.09$
10	$86 \pm 5$	$7.2 \pm 0.3$	$23.6 \pm 0.9$
11	$63 \pm 1$	$1.11 \pm 0.05$	$1.48 \pm 0.05$

# Comparison between $\gamma$ -spectroscopy and ICP-MS measurements

		$^{40}\text{K}$ [mBq/kg]	$^{232}\text{Th}$ [mBq/kg]	$^{238}\text{U}$ [mBq/kg]
NAC-2	$\gamma$ -spectroscopy	$81 \pm 19$	$5.0 \pm 2.0$	$22 \pm 2$
	ICP-MS	$86 \pm 5$	$7.2 \pm 0.3$	$23.6 \pm 0.9$

ICP-MS technique allows direct determination of Th and U concentration both in the whole sample and in the individual layers. Concerning  $\gamma$ -spectroscopy,  $^{232}\text{Th}$  and  $^{238}\text{U}$  are determined measuring respectively  $^{228}\text{Th}$  and  $^{226}\text{Ra}$ .



The 2 techniques are therefore complementary and in this case the results are in good agreement

# Conclusions

- Developed a reliable methodic for evaluating the K, Pb, Th and U contamination in polymeric substrates with and without metallic cladding by means of ICP-MS;
- Polymeric substrates w/o metallic cladding:
  - ✓ PEN samples are significantly cleaner in Th and U of about two orders of magnitude than PET ones.
  - ✓ The PEN sample that underwent the complete Cu and Cr deposition and PCB processing (sample 3) has an acceptable Th and U concentration. The specific activity of  $^{232}\text{Th}$  and  $^{238}\text{U}$  is compatible with the specification for cables surrounding crystals to match the rate of  $10^{-3}$  count/(kg·keV·y). [K.Kroninger et al. Technical Report GSTR-05-019];  $^{232}\text{Th}$  is in the worst case, an order of magnitude below the limit determined by MC (1.7 mBq/kg), while  $^{238}\text{U}$  is just at the limit (1.7 mBq/kg).
- Super-insulation materials:
  - ✓ Sample 9 ("COOL CAT 2" from Air Liquid) and sample 11 (Aluminized Teflon, FEP) are cleaner than sample 10 (NAC-2). Contamination level in Th and U for samples 9 and 11 are roughly the same and the specific activity of  $^{232}\text{Th}$  is well below (about a factor 5) the limit derived from the MC (5 mBq/kg).

# $\text{GeO}_2$ powder isotopic analysis

- Sample 1, Ge 1711, P8. & Sample 2, Ge 1714, P8:  
200 mg of each sample dissolved in solution of nitric acid, hydrofluoric acid, water (1:1:1). The obtained solution has been diluted up to about 1 ppm of Ge before measurement.

	$^{70}\text{Ge}$ [ % ]	$^{72}\text{Ge}$ [ % ]	$^{73}\text{Ge}$ [ % ]	$^{74}\text{Ge}$ [ % ]	$^{76}\text{Ge}$ [ % ]
<i>S 1 Ge 1711</i>	$0.0014 \pm 0.0001$	$0.0273 \pm 0.0005$	$0.110 \pm 0.001$	$10.35 \pm 0.02$	<b><math>89.51 \pm 0.02</math></b>
<i>S 2 Ge 1714</i>	$<0.0006$	$0.028 \pm 0.002$	$0.129 \pm 0.004$	$10.94 \pm 0.02$	<b><math>88.90 \pm 0.02</math></b>

The uncertainties reported in the table are statistical (1s). The total error of result, evaluated for comparison with previous measurements, is  $<1$  on  $^{76}\text{Ge}$  isotope percentage abundance.

# Ge isotopic measurement for GNO

	Ge70 %	Ge72 %	Ge73 %	Ge74 %	Ge76 %
<b>Carrier 70Ge</b>	<b>96,47</b> <b>96,784</b>	<b>1,14</b> <b>1,090</b>	<b>0,35</b> <b>0,320</b>	<b>1,46</b> <b>1,316</b>	<b>0,57</b> <b>0,490</b>
<b>Carrier 72Ge</b>	<b>0,36</b> <b>0,355</b>	<b>98,54</b> <b>98,609</b>	<b>0,23</b> <b>0,223</b>	<b>0,73</b> <b>0,683</b>	<b>0,14</b> <b>0,129</b>
<b>Carrier 74Ge</b>	<b>0,42</b> <b>0,444</b>	<b>0,64</b> <b>0,639</b>	<b>0,23</b> <b>0,230</b>	<b>98,5</b> <b>98,486</b>	<b>0,21</b> <b>0,201</b>

BLUE VALUE=LNGS measure carried out by ICP-MS  
(quadrupole mass analyzer)

GREEN VALUE=MC-ICP-MS (double focusing-  
multicollector) carried out from ITU of JRC Karlsruhe