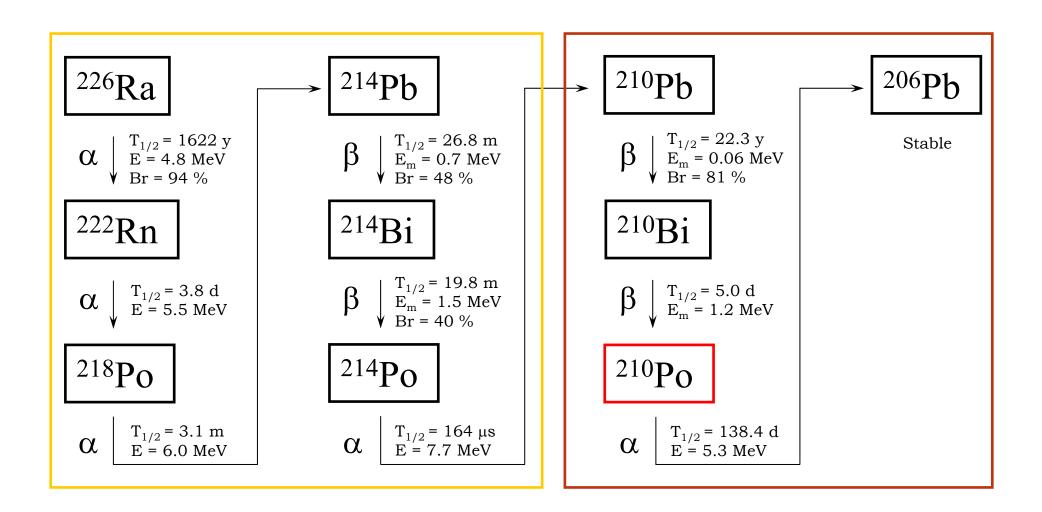
## <sup>222</sup>Rn daughters on the copper surface

M. Wojcik, G. Zuzel

- Testing cleaning procedures
  - etching
  - electropolishing
- □ Comparing etching with electropolishing
- Conclusions

## Why to investigate <sup>222</sup>Rn daughters?



- Screening of  $^{210}$ Po with an alpha spectrometer 50 mm Si-detector, bcg  $\sim 5 \, \alpha/d \, (1-10 \, MeV)$  sensitivity  $\sim 20 \, mBq/m^2 \, (100 \, mBq/kg, \, ^{210}$ Po)
- Screening of <sup>210</sup>Bi with a beta spectrometer
   2×50 mm Si(Li)-detectors, bcg ~ 0.18/0.40 cpm sensitivity ~ 10 Bq/kg (<sup>210</sup>Bi)
- Screening of <sup>210</sup>Pb (46.6 keV line) with a gamma spectrometer
   25 % n-type HPGe detector with an active and a passive shield, sensitivity ~ 20 Bq/kg
- □ Only small samples can be handled artificial contamination needed: copper discs loaded with <sup>222</sup>Rn daughters

- □ LENS electrolytic copper used to fabricate sample discs (50 mm diameter, 1 mm thickness)
- □ Discs cleaned applying "Majorana procedure" (5 min in 1% H<sub>2</sub>SO<sub>4</sub> + 3% H<sub>2</sub>O<sub>2</sub>; 5 min in 1% citric acid; rinsing with distilled water)
- □ Discs placed for 4 months in a strong <sup>222</sup>Rn source (1.4 MBq)

Discs before and after cleaning



Discs loaded with <sup>222</sup>Rn daughters



- □ Testing cleaning procedures
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- "Majorana procedure" tested:

  copper discs 5 min in 1% H<sub>2</sub>SO<sub>4</sub> + 3% H<sub>2</sub>O<sub>2</sub>

  5 min in 1% citric acid

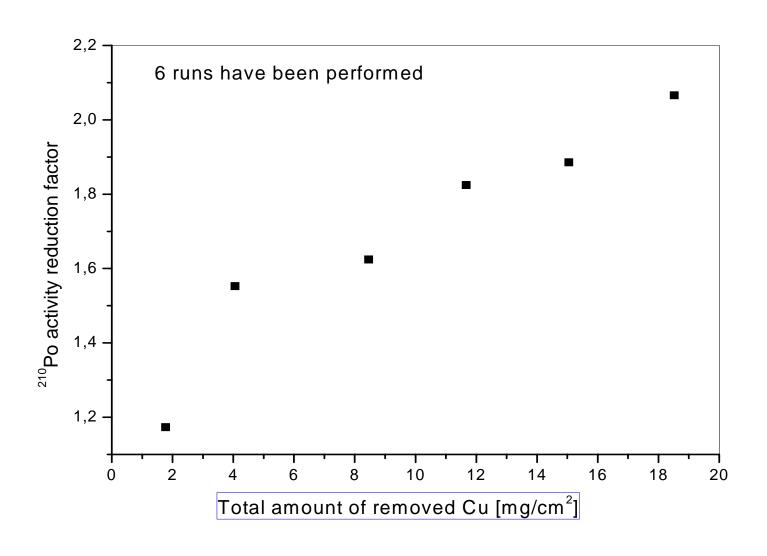
  rinsing with water
- Solution volume: each time 250 cm<sup>3</sup>
- Temperature: 20 °C
- <sup>210</sup>Po measured on both disc sides
- Blank signal:  $(0.0042 \pm 0.0005)$  cpm
- Disc loaded with <sup>210</sup>Po: side a:  $(2.97 \pm 0.03)$  cpm side b:  $(2.64 \pm 0.03)$  cpm

Results:

Loaded disc, side a:  $(2.97 \pm 0.03)$  cpm

side b:  $(2.64 \pm 0.03)$  cpm

Run No.	Disc side	<sup>210</sup> Po activity [cpm]	<sup>210</sup> Po reduction factor R	Amount of removed Cu	Remarks
1	a	$2.36 \pm 0.03$	1.3	$(1.77 \pm 0.02) \text{ mg/cm}^2$	Acid mixed during etching
1	b	$2.16 \pm 0.02$	1.2	1.98 µm	
2	a	$1.83 \pm 0.04$	1.3	$(2.29 \pm 0.02) \text{ mg/cm}^2$	Acid mixed during etching
	b	$1.79 \pm 0.03$	1.2	2.56 μm	
3	a	$1.84 \pm 0.03$	0.99	$(4.40 \pm 0.02) \text{ mg/cm}^2$	Acid mixed during etching
	b	$1.62 \pm 0.03$	1.1	4.91 μm	
4	a	$1.65 \pm 0.03$	1.1	$(3.21 \pm 0.02) \text{ mg/cm}^2$	Acid mixed during etching
	b	$1.43 \pm 0.02$	1.1	3.58 µm	
5	a	$1.62 \pm 0.03$	1.0	$(3.38 \pm 0.02) \text{ mg/cm}^2$	Acid mixed during etching
	b	$1.35 \pm 0.02$	1.1	3.77 μm	
6	a	$1.47 \pm 0.02$	1.1	$(3.47 \pm 0.02) \text{ mg/cm}^2$	Acid mixed during etching
U	b	$1.25 \pm 0.03$	1.1	3.87 µm	



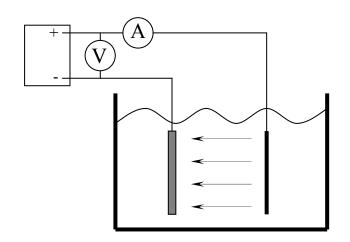
### Results for <sup>210</sup>Pb, <sup>210</sup>Bi and <sup>210</sup>Po:

Isotope	Original activity [cpm]	Activity after cleaning [cpm]	Reduction factor R	Amount of removed Cu	Remarks
<sup>210</sup> Pb	$1.49 \pm 0.04$	< 0.022	> 68	2.01	Only side a was investigated
<sup>210</sup> Bi	$31.17 \pm 0.71$	$0.77 \pm 0.02$	40.5	3.91 mg/cm <sup>2</sup> 4.4 μm	Only side a was investigated
<sup>210</sup> Po	$2.55 \pm 0.01$	$2.06 \pm 0.01$	1.2		Only side a was investigated

- □ Technique
- □ Testing cleaning procedures
  - etching
  - electropolishing
- □ Comparing etching with electropolishing
- Conclusions

## Electropolishing, disc No. 4

- □ Both disc sides investigated separately for <sup>210</sup>Po
- □ Electrolyte:  $85 \% H_3PO_4 + 5 \% 1$ -butanol ( $C_4H_{10}O$ )
- Only one cathode (Cu disc)
- □ Several runs performed, each time using a new cathode and fresh solution



$$U = 1.8 \text{ V}$$
  
 $I = 150 - 10 \text{ mA}$ 

# Electropolishing, disc No. 4

Loaded disc side a:  $(9.52 \pm 0.06)$  cpm

side b:  $(1.78 \pm 0.04)$  cpm

Run No.	Disc side	<sup>210</sup> Po activity [cpm]	<sup>210</sup> Po reduction factor R	Amount of removed Cu*	Remarks
1	a	$0.50 \pm 0.03$	19		Polished for 35 min Total charge: 70 mAh  Polished for 35 min
1	b	$1.38 \pm 0.03$	1.3		
2	a	$0.062 \pm 0.003$	8	17 mg/cm <sup>2</sup>	
	b	$0.74 \pm 0.01$	1.9		Total charge: 70 mAh
3	a	$0.024 \pm 0.002$	2.6		Polished for 35 min
	b	$0.017 \pm 0.002$	44		Total charge: 70 mAh

<sup>-</sup> disc side facing the cathode

Results:

After	a	$0.024 \pm 0.002$	397	≤ 12.6 µm	Polished for 70 min
all	b	$0.017 \pm 0.002$	105	≤ 6.3 µm	Polished for 35 min

<sup>\*)</sup> measured after all runs

## Electropolishing, disc No. 3

- □ Both sides investigated separately for <sup>210</sup>Po
- □ Electrolyte:  $85 \% H_3PO_4 + 5 \% 1$ -butanol ( $C_4H_{10}O$ )
- □ Only one cathode (Cu disc)
- □ One run performed, disc was turned around several times
- □ Total polishing time: 3 h

#### Results:

Disc side	Original <sup>210</sup> Po activity [cpm]	<sup>210</sup> Po activity after pol. [cpm]	<sup>210</sup> Po reduction factor R	Amount of removed Cu	Remarks
a	$2.18 \pm 0.02$	$0.011 \pm 0.001$	198	20 mg/cm <sup>2</sup>	Facing the cathode 3 times, each time for 30 min
b	$2.45 \pm 0.03$	$0.014 \pm 0.001$	175	22.3 μm	Facing the cathode 3 times, each time for 30 min

- □ Testing cleaning procedures
  - etching
  - electropolishing
- □ Comparing etching with electropolishing
- Conclusions

### Comparing etching with electropolishing

- □ Amount of removed material:
  - after 6 "Majorana" runs (30 min): 18.5 mg/cm<sup>2</sup>
  - after one polishing run (35 min): 5.7 mg/cm<sup>2</sup>
- □ Amount of removed <sup>210</sup>Po activity:
  - after 6 "Majorana" runs (35 min, 18.5 mg/cm<sup>2</sup>):  $R_{av} = 2.1$
  - after long-polishing run (3 h, 20 mg/cm<sup>2</sup>):  $R_{av} = 187$
- Amount of removed <sup>210</sup>Pb and <sup>210</sup>Bi activity:
  - one "Majorana" run (5 min, ~3 mg/cm<sup>2</sup>): already effective
  - electropolishing: not yet tested, but should work as well

#### Conclusions

- □ Etching and electropolishing remove up to 20 mg/cm<sup>2</sup> Cu (depending on the exposure time)
- □ 210Po deposited on- or close to the copper discs surface (relatively narrow α-peaks)
- □ Etching does not remove <sup>210</sup>Po from the copper probably re-deposition (peaks shape)
- □ Long electropolishing reduces <sup>210</sup>Po activity by a factor of ~200 much more effective than etching
- □ Etching removes most of <sup>210</sup>Pb and <sup>210</sup>Bi (> 98 %)
- □ Ongoing tests: <sup>210</sup>Pb/<sup>210</sup>Bi removal by electropolishing